

CLASS TEST

GATE
exclusive

S.No. : 05 GH1_ME_GS_240819

Material Science



MADE EASY

India's Best Institute for IES, GATE & PSUs

Delhi | Noida | Bhopal | Hyderabad | Jaipur | Lucknow | Indore | Pune | Bhubaneswar | Kolkata | Patna

Web: www.madeeasy.in | E-mail: info@madeeasy.in | Ph: 011-45124612

CLASS TEST 2019-2020

MECHANICAL ENGINEERING

Date of Test : 24/08/2019

ANSWER KEY ➤ Material Science

- | | | | | |
|--------|---------|---------|---------|---------|
| 1. (b) | 7. (c) | 13. (b) | 19. (a) | 25. (c) |
| 2. (c) | 8. (a) | 14. (d) | 20. (a) | 26. (b) |
| 3. (c) | 9. (c) | 15. (c) | 21. (d) | 27. (b) |
| 4. (a) | 10. (b) | 16. (d) | 22. (b) | 28. (b) |
| 5. (b) | 11. (b) | 17. (d) | 23. (c) | 29. (a) |
| 6. (c) | 12. (b) | 18. (a) | 24. (c) | 30. (a) |

Detailed Explanations

1. (b)

$$\sqrt{3}a = 4r \rightarrow \text{BCC}$$

$$\sqrt{2}a = 4r \rightarrow \text{FCC}$$

$$a = 2r \rightarrow \text{HCP}$$

3. (c)

Silicon is used for promoting graphitization in cast iron.

4. (a)

Tungsten \rightarrow BCC structure

Atomic packing factor = 0.68

5. (b)

7 – Crystal systems

14 – Bravais lattices

6. (c)

Cold working – when material is deformed below its recrystallization temperature.

8. (a)

Peritectic reaction : Liquid A + Solid B \rightarrow Solid C.

9. (c)

$$\text{Atomic packing factor} = \frac{\text{Volume of atoms in a unit cell}}{\text{Volume of unit cell}}$$

$$= \frac{4 \times \frac{4}{3}\pi r^3}{a^3}$$

$$\therefore r = \left(\frac{a}{2\sqrt{2}} \right) \quad \dots(1)$$

$$= \frac{4 \times \frac{4}{3}\pi \left(\frac{a}{2\sqrt{2}} \right)^3}{a^3} = 0.74$$

10. (b)

$$\text{Body centered cubic} = \frac{1}{8} \times 8 + 1 = 2$$

13. (b)

$$n_{\text{HCP}} = 6$$

$$n_{\text{DC}} = 8$$

$$\frac{n_{\text{HCP}}}{n_{\text{DC}}} = \frac{6}{8} = 0.75$$

17. (d)

Length of the edge of unit cell, $l = 654 \text{ pm} = 6.54 \times 10^{-8} \text{ cm}$
 Volume, $V = l^3 = (6.54 \times 10^{-8})^3 \text{ cm}^3$

Molecular mass of KBr = 119 g/mol

Density, $\rho = 2.75 \text{ g/cm}^3$

$$\rho = \frac{n \times M}{V \times N_A}$$

$$\Rightarrow n = \frac{\rho \times V \times N_A}{M}$$

$$\Rightarrow n = \frac{2.75 \times (6.54)^3 \times 10^{-24} \times (6.023) \times 10^{23}}{119}$$

$$\Rightarrow n = 3.89 \simeq 4 \quad (\text{Since number of atoms will be whole number})$$

19. (a)

$$\sigma = k\epsilon^n$$

$$\sigma = 800\epsilon^{0.3}$$

$$n = \epsilon$$

$$\sigma = 800(0.3)^{0.3}$$

$$= 557.476 \text{ MPa} \quad \dots(1)$$

$$\sigma = \sigma_0(1 + e)$$

$$\dots(2)$$

σ_0 = Ultimate tensile strength, e = engineering strain

$$\epsilon = \ln(1 + e)$$

$$1 + e = e^\epsilon = e^{0.3} = 1.349$$

From Eqs. (1) and (2), we get

$$557.476 = \sigma_0(1.349)$$

$$\sigma_0 = 413.25 \text{ MPa}$$

20. (a)

$$\begin{aligned} BHN &= \frac{2P}{\pi D [D - \sqrt{D^2 - d^2}]} = \frac{2 \times 500}{\pi \times 10 [10 - \sqrt{10^2 - 1.64^2}]} \\ &= 235 \end{aligned}$$

22. (b)

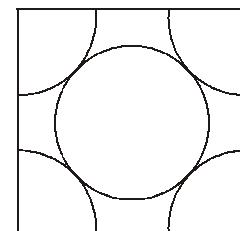
FCC (100) plane

$$\text{Number of atoms} = \frac{1}{4} \times 4 + 1 = 2$$

$$\text{Area} = a \times a = a^2$$

$$\therefore \text{Planar density} = \left(\frac{2}{a^2} \right)$$

$$\Rightarrow x = 2$$

**23. (c)**

Muntz metal - Cu and Zn

Inconel - Cr, Ni and Fe

Monel - Ni and Cu

Invar - Ni and Fe

24. (c)

$$E_c = 0.2(300) + (1 - 0.2)100 \\ = 60 + 80 = 140 \text{ GPa}$$

$$\text{Load fraction, } \frac{P_f}{P_m} = \frac{0.2(300)}{0.8(100)} = 0.75$$

26. (b)

Interplaner spacing

$$d_{hkl} = 0.1013 \text{ nm}$$

Order of reflection, $n = 1$

Wavelength, $\lambda = 0.1790 \text{ nm}$

$$\sin\theta = \frac{n\lambda}{2d_{hkl}} = \frac{1 \times 0.1790}{2 \times 0.1013} = 0.884$$

\Rightarrow

$$\theta = \sin^{-1}(0.8841) = 62.13^\circ$$

Diffraction angle = $2\theta = 124.26^\circ$

27. (b)

$$N_v = N \times e^{\left(\frac{-Q_v}{kT}\right)}$$

$$N = 8 \times 10^{28} \text{ atoms/m}^3$$

$$Q_v = 0.9 \text{ eV}$$

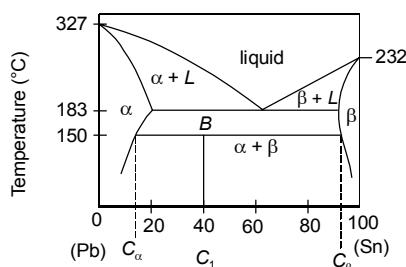
$$k = 8.62 \times 10^{-5} \text{ eV/K}$$

$$T = 1000 + 273 = 1273 \text{ K}$$

Putting the values,

$$N_v = 2.2 \times 10^{25} \text{ vacancies/m}^3$$

28. (b)



Composition of β -phase is tin nearly 98% and lead 2% at 150°C .

29. (a)

Let p, q, r be the intercepts on x, y and z axes respectively, then,

$$p : q : r = \frac{a}{h} : \frac{b}{k} : \frac{c}{l}$$

where a, b and c are primitives and h, k and l are miller indices.

Given:

$$a = 1.2 \text{ \AA}$$

$$b = 1.8 \text{ \AA}$$

$$c = 2 \text{ \AA}$$

$$h = 2$$

$$k = 3, l = 1$$

$$p : q : r = \frac{1.2}{2} : \frac{1.8}{3} : \frac{2}{1} = 0.6 : 0.6 : 2$$

$$p : q = 0.6 : 0.6$$

$$\Rightarrow 1.2 : q = 0.6 : 0.6$$

$$q = \frac{1.2 \times 0.6}{0.6} = 1.2 \text{ \AA}$$

Similarly,

$$q : r = 0.6 : 2$$

$$1.2 : r = 0.6 : 2$$

$$r = \frac{1.2 \times 2}{0.6} = 4 \text{ \AA}$$

$$\therefore \frac{q}{r} = \frac{1.2}{4} = 0.3$$

30. (a)

$$\sigma_T = \sigma_o + K\varepsilon_T^n \quad \dots(i)$$

$$\begin{aligned} \Rightarrow \frac{d\sigma_T}{d\varepsilon_T} &= 0 + K \cdot n \cdot \varepsilon_T^{n-1} \\ &= Kn\varepsilon_T^{n-1} \end{aligned} \quad \dots(ii)$$

From equation (i)

$$K = \frac{\sigma_T - \sigma_o}{(\varepsilon_T)^n}$$

Substituting the above value in equation (ii)

$$\begin{aligned} \frac{d\sigma_T}{d\varepsilon_T} &= \left(\frac{\sigma_T - \sigma_o}{\varepsilon_T^n} \right) (n \varepsilon_T^{(n-1)}) = \frac{(\sigma_T - \sigma_o)n}{\varepsilon_T} \\ &= \frac{(300 - 200) \times 0.3}{0.05} = 600 \text{ MPa} \end{aligned}$$

