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# ENVIRONMENT ENGINEERING

## CIVIL ENGINEERING

Date of Test : 14/08/2022

### ANSWER KEY >

- |        |         |         |         |         |
|--------|---------|---------|---------|---------|
| 1. (c) | 7. (c)  | 13. (b) | 19. (a) | 25. (a) |
| 2. (c) | 8. (b)  | 14. (d) | 20. (d) | 26. (c) |
| 3. (d) | 9. (a)  | 15. (d) | 21. (b) | 27. (c) |
| 4. (a) | 10. (c) | 16. (a) | 22. (a) | 28. (c) |
| 5. (b) | 11. (d) | 17. (c) | 23. (b) | 29. (a) |
| 6. (d) | 12. (b) | 18. (c) | 24. (c) | 30. (d) |

## DETAILED EXPLANATIONS

1. (c)

$$\begin{aligned} \text{BOD}_5 &= (\text{Initial DO} - \text{Final DO}) \times \text{Dilution factor} \\ &= (9.36 - 6.2) \times \frac{100}{2.5} = 126.4 \text{ mg/l} \end{aligned}$$

4. (a)

$$L = \frac{(0.5 \text{ m}^3/\text{s}) \times (86400 \text{ s/d})}{150 \text{ m}^3/\text{day/m}} = 288 \text{ m}$$

5. (b)

$$\begin{aligned} (100 - P_1) V_1 &= (100 - P_2) V_2 \\ \Rightarrow \frac{V_2}{V_1} &= \frac{100 - P_1}{100 - P_2} \\ &= \frac{100 - 97}{100 - 95} = \frac{3}{5} = 0.6 \end{aligned}$$

So, decrease in volume by 40%.

7. (c)

The drop manholes are provided in sewerage system when there is change in elevation of ground level.

10. (c)

$$\text{Sound pressure level (dB)} = 20 \log_{10} \left( \frac{P}{P_0} \right)$$

where P is sound pressure in N/m<sup>2</sup>

and P<sub>0</sub> is reference pressure (= 2 × 10<sup>-5</sup> N/m<sup>2</sup> or 20 μPa)

For given sound pressure, P = 3000 μ bar = 300 N/m<sup>2</sup> [ ∵ 1 bar = 1 × 10<sup>5</sup> Pa = 1 × 10<sup>5</sup> N/m<sup>2</sup> ]

$$\therefore \text{Sound pressure level} = 20 \log_{10} \left( \frac{300}{2 \times 10^{-5}} \right) = 143.52 \text{ dB}$$

11. (d)

**Statement-1** : Raw sludge is more objectionable in comparison to secondary sludge because it has more putrescible organic matter.

**Statement-3** : Secondary sludge from trickling filter has solid content (2-4)% and activated sludge process has solid content of (1-2)%.

12. (b)

For SO<sub>2</sub>

$$\text{Minimum height of chimney} = 14 (Q_s)^{1/3} \text{ m}$$

$$Q_s = \text{SO}_2 \text{ emission in kg/hr}$$

$$Q_s = 0.72 \text{ tonnes/day}$$

$$= 720 \text{ kg/24 hr}$$

$$= 30 \text{ kg/hr}$$

$$\text{So, Height of chimney} = 14 (30)^{1/3} \\ = 43.5 \text{ m}$$

13. (b)

Statement 2 and 3 are the properties of iron salts not alum salts.

15. (d)

Velocity of sand particles,

$$V'_s = \frac{g}{18} (G - 1) \frac{d^2}{\nu}$$

$$\Rightarrow V'_s = \frac{9.81}{18} \times (2.65 - 1) \times \frac{(0.01 \times 10^{-3})^2}{10^{-3} / 1000}$$

$$\Rightarrow V'_s = \frac{9.81}{18} \times 1.65 \times 10^{-4}$$

$$\Rightarrow V'_s = 8.9925 \times 10^{-5} \text{ m/s}$$

$$\text{Setting time} = \frac{\text{Volume of tank}}{\text{Water flow rate}}$$

$$= \frac{900 \times 3}{8000} \text{ days} = \frac{2700}{8000} \times 24 = 8.1 \text{ hours}$$

$$\text{Settling velocity, } V_s = \frac{\text{Depth of tank}}{\text{Settling time}}$$

$$= \frac{3}{8.1 \times 3600} = 1.0288 \times 10^{-4} \text{ m/s}$$

$$\therefore \text{Removal efficiency} = \frac{V'_s}{V_s} \times 100$$

$$= \frac{8.9925 \times 10^{-5}}{1.0288 \times 10^{-4}} \times 100 = 87.41\%$$

16. (a)

$$\text{Recirculation factor, } F = \frac{1 + R/I}{\left(1 + \frac{0.1R}{I}\right)^2} = \frac{1 + 1.5}{(1 + 0.1 \times 1.5)^2} = 1.89$$

$$\text{Efficiency of filter, } \eta = \frac{100}{1 + 0.0044 \sqrt{\frac{Y}{V \times F}}} \quad (\text{where } V = 0.12 \text{ ha.m})$$

$$\Rightarrow 80 = \frac{100}{1 + 0.0044 \sqrt{\frac{Y}{0.12 \times 1.89}}}$$

$$\Rightarrow Y = 732.18 \text{ kg/day}$$

17. (c)

Velocity gradient,  $G = \sqrt{\frac{P}{\mu V}}$

Power input per unit volume,  $P = \mu G^2 V$

$$= 1 \times 10^{-3} \times (450)^2 \times 4$$

$$= 810 \text{ W}$$

Now, power input per unit volume =  $\frac{P}{V} = \frac{810}{4} = 202.5 \text{ W/m}^3$

18. (c)

Sound pressure,  $L = 20 \log_{10} \left( \frac{P_{\text{rms}}}{20 \mu\text{Pa}} \right) \text{ dB}$

$$L = 20 \log_{10} \frac{2000}{20}$$

$$= 20 \log_{10} 100$$

$$= 40 \text{ dB}$$

So, resultant of two sound sources of 40 dB and 40 dB = 43 dB

19. (a)

For Infiltration gallery,

$$Q = kL \left( \frac{H^2 - h^2}{2R} \right)$$

where,

$$Q = 600 \text{ m}^3/\text{day}$$

$$H = 6 \text{ m}, k = 100 \text{ m/day}$$

$$R = 100 \text{ m}$$

$$h = 6 - 4 = 2 \text{ m}$$

$$\therefore 600 \text{ m}^3/\text{day} = 100 \text{ m/d} \times L \text{ (in m)} \times \left( \frac{6^2 - 2^2}{2 \times 100} \right) \frac{\text{m}^2}{\text{m}}$$

$$\Rightarrow 600 = \left( 100 \times L \times \frac{32}{200} \right)$$

$$\Rightarrow L = 37.5 \text{ m}$$

21. (b)

$$\begin{aligned} \text{Required chlorine dose} &= \text{Chlorine demand} + \text{Residual chlorine} \\ &= 1.28 + 0.22 = 1.5 \text{ mg/l} \end{aligned}$$

$$\begin{aligned} \therefore \text{Total chlorine required per day} \\ &= 1.5 \times 10^{-6} \times 6.4 \times 10^6 = 9.6 \text{ kg} \end{aligned}$$

Available chlorine in bleaching powder = 20%

$$\begin{aligned} \therefore \text{Bleaching powder required} \\ &= \frac{9.6}{0.20} = 48 \text{ kg} \end{aligned}$$

22. (a)

$$N = N_0 10^{kt} \quad (\text{1st order reaction})$$

$$\therefore 10^9 = 10^3 (10)^{k \times 10}$$

$$\Rightarrow 10^6 = (10)^{10k}$$

$$\Rightarrow k = 0.6$$

$$\text{Now, } N = 2N_0$$

$$\therefore 2 = 10^{0.6t}$$

$$\Rightarrow t = 0.5 \text{ hr}$$

23. (b)

$$\text{Sewage produced} = 400000 \text{ l/day}$$

$$5\text{-day BOD of sewage} = 140 \text{ mg/l} \quad (\because \text{When water is solvent, } 1 \text{ ppm} = 1 \text{ mg/l})$$

$$\text{BOD of effluent} = 15 \text{ ppm} = 15 \text{ mg/l}$$

$$\therefore \text{BOD removed by pond} = (140 - 15) = 125 \text{ ppm} = 125 \text{ mg/l}$$

$$\begin{aligned} \therefore \text{Sewage produced per day} \\ &= 400000 \times 125 \\ &= 50 \times 10^6 \text{ mg} = 50 \text{ kg} \end{aligned}$$

$$\text{Organic loading rate} = 25 \text{ kg/ha/day}$$

$$\therefore \text{Required area} = \frac{\text{Sewage produced}}{\text{Organic loading rate}}$$

$$= \frac{50}{25} = 2 \text{ ha}$$

24. (c)

Hardness is due to multivalent cations.

$$\begin{aligned} \therefore \text{Total hardness} &= [\text{Mg}^{+2}] \times \frac{\text{Eq. weight of CaCO}_3}{\text{Eq. weight of Mg}^{2+}} + [\text{Ca}^{+2}] \times \frac{\text{Eq. weight of CaCO}_3}{\text{Eq. weight of Ca}^{2+}} \\ &+ [\text{Al}^{3+}] \times \frac{\text{Eq. weight of CaCO}_3}{\text{Eq. weight of Al}^{3+}} \\ &= 48 \times \frac{50}{12} + 45 \times \frac{50}{20} + 5 \times \frac{50}{9} = 340.28 \text{ mg/l as CaCO}_3 \end{aligned}$$

25. (a)

Total no. of cigarettes consumed per hour =  $50 \times 1.3 = 65$

Total formaldehyde produced per hour =  $65 \times 1.3 = 84.5 \text{ mg}$

This 84.5 mg of formaldehyde provide per hour gets mixed with  $1300 \text{ m}^3/\text{hr}$  of fresh air injected into the room.

$$\text{Hence concentration of formaldehyde in the room air} = \frac{84.5}{1300} = 0.065 \text{ mg/m}^3$$

26. (c)

Flow rate,  $Q_0 = 800 \text{ m}^3/\text{hr}$

Influent BOD,  $S_0 = 125 \text{ mg/l}$

Effluent BOD,  $S = 10 \text{ mg/l}$

Hydraulic retention time =  $7.5 \text{ hr} = 0.3125 \text{ days}$

Mean-cell residence time, HRT  $\theta_c = 300 \text{ hr} = \frac{300}{24} \text{ days} = 12.5 \text{ days}$

MLSS (X) =  $3000 \text{ mg/l}$

Volume =  $500 \text{ m}^3$

Food (F) =  $Q_0 S_0$

$$= \left( 800 \frac{\text{m}^3}{\text{hr}} \right) \left( 125 \frac{\text{mg}}{\text{l}} \right) = 2400 \text{ kg/day}$$

Biomass (M) =  $VX$

$$= \left( 500 \text{ m}^3 \right) \left( 3000 \frac{\text{mg}}{\text{l}} \right)$$

$$= 1500 \text{ kg}$$

$$\therefore \frac{F}{M} = \frac{Q_0 S_0}{VX} = \frac{2400}{1500}$$

$$= 1.6 \text{ kg BOD day/kg MLSS}$$

27. (c)

From place A to B,

$$\text{Ambient lapse rate} = \frac{21 - 18.5}{350 - 45} \times 1000 = 8.197 < 9.8^\circ\text{C}/\text{km}$$

Since lapse rate is less than the adiabatic lapse rate, so ambient lapse rate is said to be subadiabatic.

From B to C,

$$\text{Ambient lapse rate} = \frac{18.5 - 14}{500 - 350} \times 1000 = 30^\circ\text{C}/\text{km} > 9.8^\circ\text{C}/\text{km}$$

So lapse rate is superadiabatic.

28. (c)

$$\text{Solid content} = 3.5\%$$

$$\therefore \text{Water content} = 100 - 3.5 = 96.5\%$$

$$\frac{100}{S_{\text{sludge}}} = \frac{96.5}{1} + \frac{3.5}{2.4}$$

$$\Rightarrow S_{\text{sludge}} = 1.02084$$

$$\begin{aligned} \therefore \rho_{\text{sludge}} &= 1.02084 \times 1000 \\ &= 1020.84 \text{ kg/m}^3 \end{aligned}$$

After sludge gets thickened,

$$V_2 = \frac{V_1}{2}$$

$$\therefore V_2 = \frac{(100 - P_1)}{(100 - P_2)} \times V_1$$

$$\Rightarrow \frac{V_1}{2} = \frac{(100 - 96.5)}{(100 - P_2)} \times V_1$$

$$\Rightarrow 100 - P_2 = 200 - 193.0$$

$$\Rightarrow 100 - P_2 = 7$$

$$\Rightarrow P_2 = 93\%$$

$$\therefore \text{Solid content} = 100 - 93 = 7\%$$

30. (d)

The field capacity of municipal solid waste is the total moisture that can be retained in a waste sample against gravity.

